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#### Key indicators

Single-crystal X-ray study  
 $T = 295\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$   
 $R$  factor = 0.050  
 $wR$  factor = 0.173  
Data-to-parameter ratio = 16.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

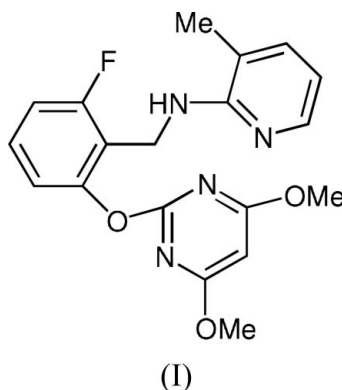
## 2-(4,6-Dimethoxy-2-pyrimidin-2-yloxy)-6-fluoro-N-(3-methylpyridyl)benzylamine

In the title compound,  $\text{C}_{19}\text{H}_{19}\text{FN}_4\text{O}_3$ , the two heterocyclic ring substituents lie on the same side of the central benzene ring.

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#### Comment

Pyrimidinyloxybenzylamine derivatives have very high weed-control activity: they are highly efficient with low toxicity, and biodegradable, safe and environmentally friendly agrochemicals (Lu *et al.*, 2001). As part of a study of their structure-activity relationship (SAR), the title compound, (I), that was a product of condensation reactions of 4,6-dimethoxy-2-methylsulfonylpyrimidine, 2-amino-3-methylpyridine and 2-fluoro-6-hydroxybenzaldehyde, was investigated.



In (I), the two heterocyclic rings lie on the same side of the benzene ring (Fig. 1); the dihedral angle between the two nitrogen-containing rings is  $85.67(10)^\circ$ . This conformation allows the formation of a centrosymmetric dimer mediated by a pair of  $\text{N}-\text{H} \cdots \text{N}$  hydrogen bonds (Table 1).

#### Experimental

2-Amino-3-methylpyridine (0.54 g, 5 mmol) was added dropwise to a methanol solution of 2-fluoro-6-hydroxybenzaldehyde (0.7 g, 5 mmol). At room temperature,  $\text{NaBH}_4$  (0.34 g) was added with stirring to give a yellow precipitate. After extraction and distillation, and drying *in vacuo*, the product was added to a flask containing a tetrahydrofuran solution (25 ml) of 4,6-dimethoxy-2-methylsulfonylpyrimidine (0.98 g, 4.5 mmol) and  $\text{K}_2\text{CO}_3$  (1.17 g) and the mixture was refluxed for 5 h. After filtration, the liquor was evaporated under vacuum to give a solid (1.38 g, yield 83%), which was recrystallized from methanol and petroleum ether (3:1) to give colorless blocks (m.p. 359–360 K).

## Crystal data

$C_{19}H_{19}FN_4O_3$   
 $M_r = 370.38$   
 Triclinic,  $P\bar{1}$   
 $a = 8.8490$  (10) Å  
 $b = 10.5810$  (10) Å  
 $c = 11.3773$  (8) Å  
 $\alpha = 113.4470$  (10)°  
 $\beta = 98.8520$  (10)°  
 $\gamma = 106.364$  (2)°  
 $V = 893.89$  (15) Å<sup>3</sup>

$Z = 2$   
 $D_x = 1.376$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 3528 reflections  
 $\theta = 3.5\text{--}27.5^\circ$   
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 295$  (1) K  
 Block, colorless  
 $0.42 \times 0.38 \times 0.22$  mm

## Data collection

Rigaku R-AXIS RAPID diffractometer  
 $\omega$  scans  
 Absorption correction: multi-scan (ABSCOR; Higashi, 1995)  
 $T_{\min} = 0.940$ ,  $T_{\max} = 0.978$   
 6292 measured reflections

3991 independent reflections  
 2983 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.022$   
 $\theta_{\text{max}} = 27.5^\circ$   
 $h = -11 \rightarrow 11$   
 $k = -12 \rightarrow 13$   
 $l = -14 \rightarrow 14$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.050$   
 $wR(F^2) = 0.173$   
 $S = 1.06$   
 3991 reflections  
 249 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.1093P)^2 + 0.0797P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.43$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.33$  e Å<sup>-3</sup>  
 Extinction correction: SHELXL97  
 Extinction coefficient: 0.050 (8)

Table 1

Hydrogen-bond geometry (Å, °).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$N1\text{--}HN1\cdots N3^i$	0.89 (2)	2.56 (2)	3.355 (2)	149 (2)

Symmetry code: (i)  $-x, -y, -z$ .

The H atom attached to atom N1 was located in a difference Fourier map and refined isotropically. Other H atoms were placed in calculated positions and allowed to ride on their parent atoms, with  $C\text{--}H = 0.98$  (aromatic), 0.97 (methylene) or 0.96 Å (methyl).  $U_{\text{iso}}(H)$  values were set at  $1.2U_{\text{eq}}(C)$  (aromatic and methylene H) or  $1.5U_{\text{eq}}(C)$  (methyl H).

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/

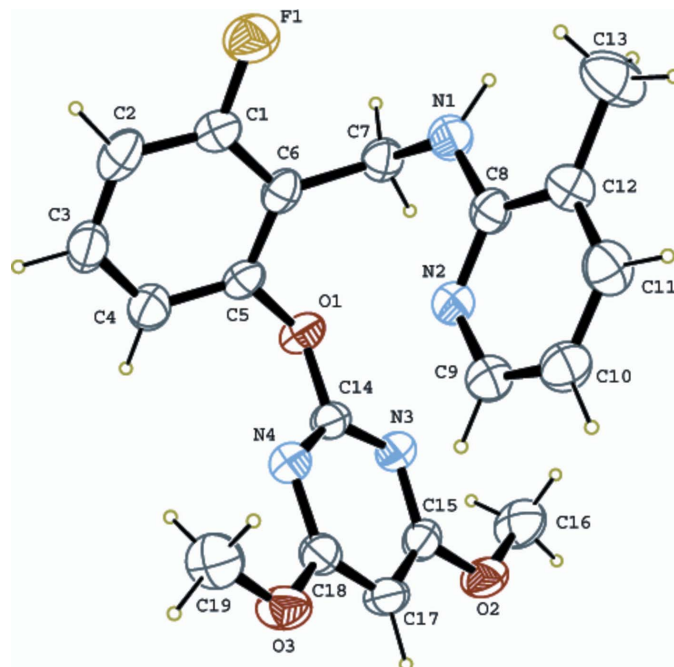


Figure 1 The structure of (I), shown with 40% probability displacement ellipsoids.

MSC, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: CrystalStructure; software used to prepare material for publication: CrystalStructure.

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